

Heat and other products induced by plasma electrolysis

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Contents

- 1) Experimental procedures;
*Material, Electrolyte, Heat calibration method
and Element analysis methods.*
- 2) Results;
Excess heat and Products.
- 3) Theoretical explanation;
Photo fission mechanism.

Electrode and Electrolyte

- Electrode;
highest quality W, (99.95% pure, supplied by High Metals Co. LTD.), dimension; $5 \times 10 \times 0.3$ mm
- Reagent: ultra high purity;
K₂CO₃, Merck Co. Ltd., metal impurity max 1ppm
- Electrolyte;
light water, purified through a milli-Q filter up to 18.3 Ohm-cm of resistivity

Cell and Electrode

- **Cell dimension;**
8 cm in diameter and 20 cm in height of Pyrex cell,
- **Cathode;**
W plate (1.0 cm x 0.5 cm) and incorporated a 1.5 ϕ , 15-cm length of W wire.
- **Anode;**
Rectangular Pt mesh had an integral lattice constructed using a 15-cm length of 0.1 cm-diameter.

Power supply and Data sampling

- DC power supply;
Takasago Products LTD, EX-1500L, 25A and 240V.
- Data processing;
Temperatures, Volt, Current → logger → computer

Electrolysis condition

- **Electrolyte;**
0.2 mol of K_2CO_3 solution.
- **Voltage;**
60 ~ 220V (plasma electrolysis)
- **Current density;**
0.3 ~ 10A/cm²
- **Electrolyte temperature;**
60 ~ 100 °C

Elements analysis

- **Sample surface;**
EDX analysis.
- **Total amount;**
ICP analysis.

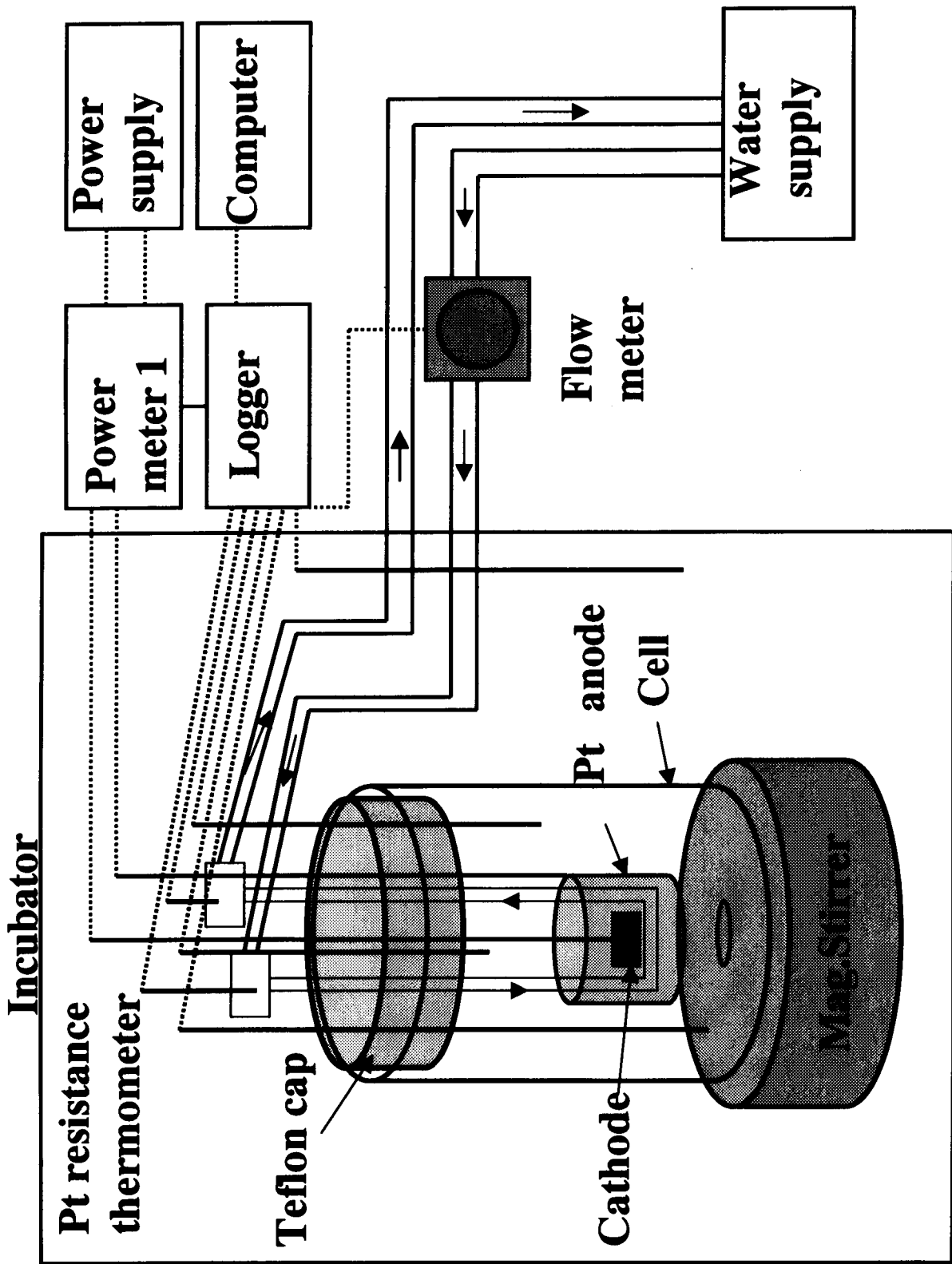
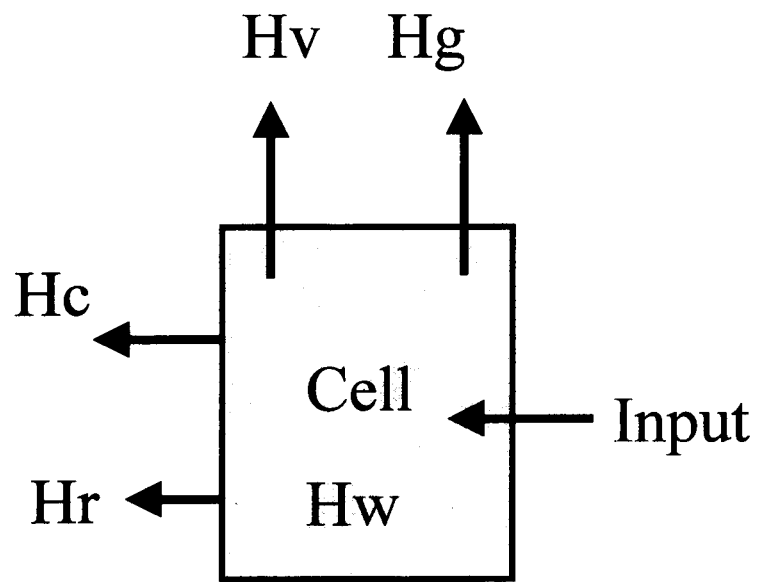


Fig.1 Experimental arrangement



Heat balance

Heat balance

$$\text{Input (W)} = I \cdot V$$

$$\text{Out} = H_g + H_w + H_c + H_r + H_v$$

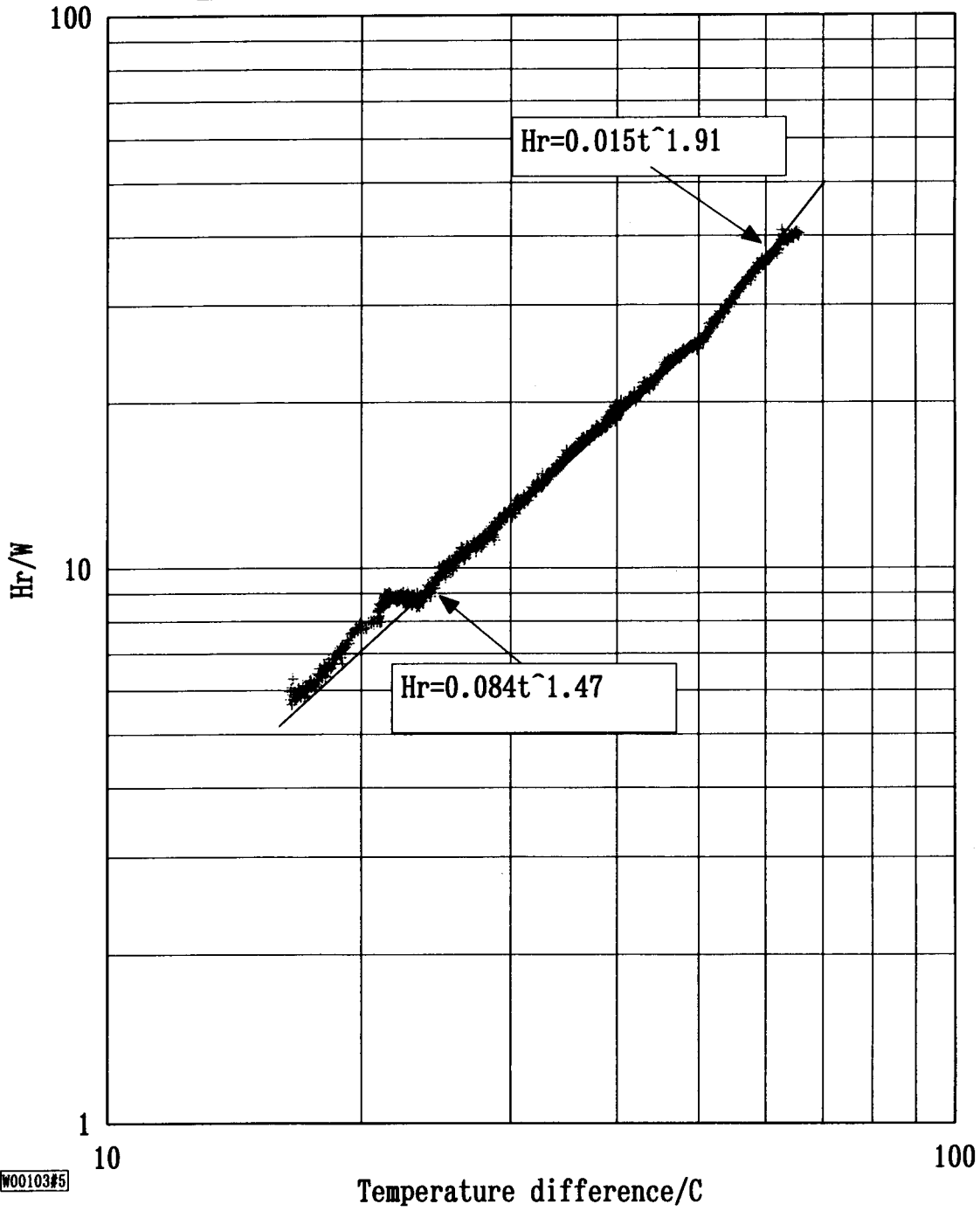
- $H_g = \text{Heat of decomposition} = 1.48 \cdot I$
- $H_w = \text{Electrolyte heat} = W_w \cdot C_w \cdot dT/t$

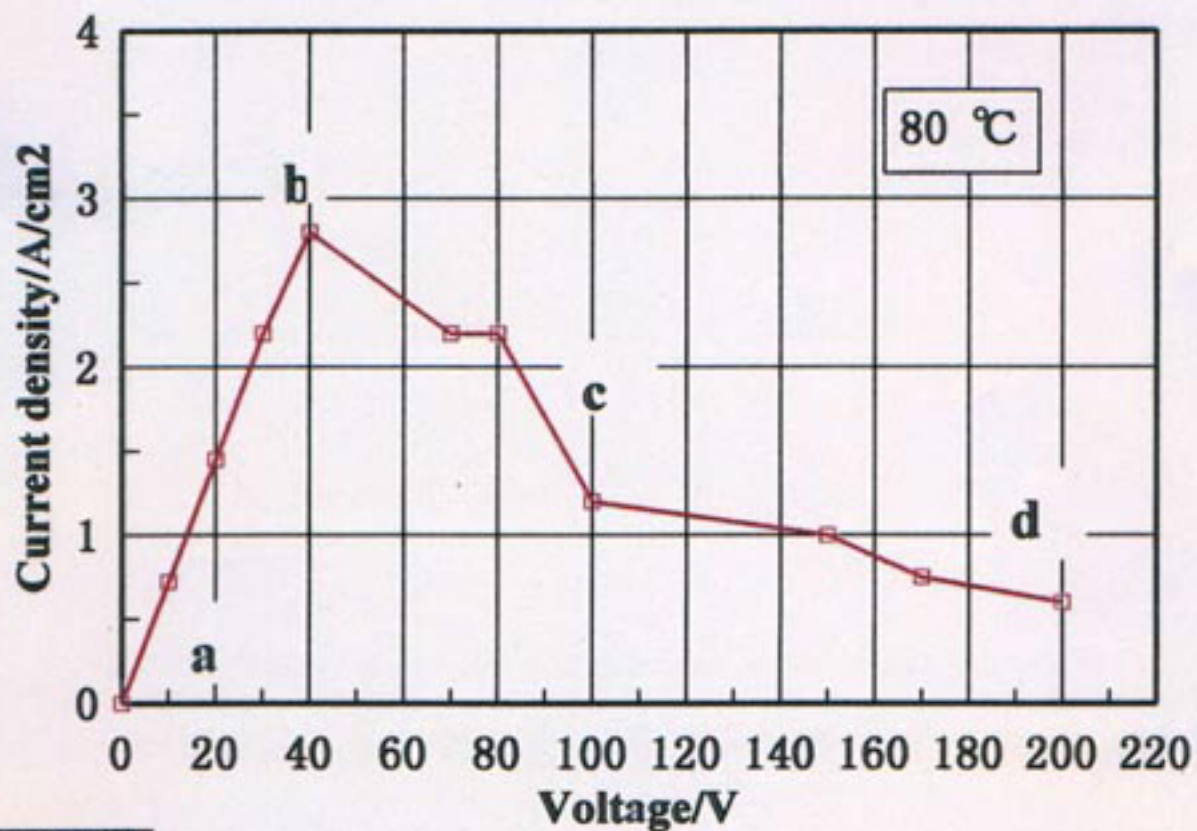
W_w : electrolyte weight, C_w : heat capacity, dT : temperature difference
 t : interval time of data sampling
- $H_c = \text{Heat of coolant} = W_c \cdot C_c \cdot dT$

W_c : coolant weight, C_c : heat capacity, dT : temperature difference
- $H_r = \text{Heat release} = W_w \cdot C_w \cdot T_r$

T_r : temperature change
- $H_v = \text{vapor} = W_v \cdot C_c$

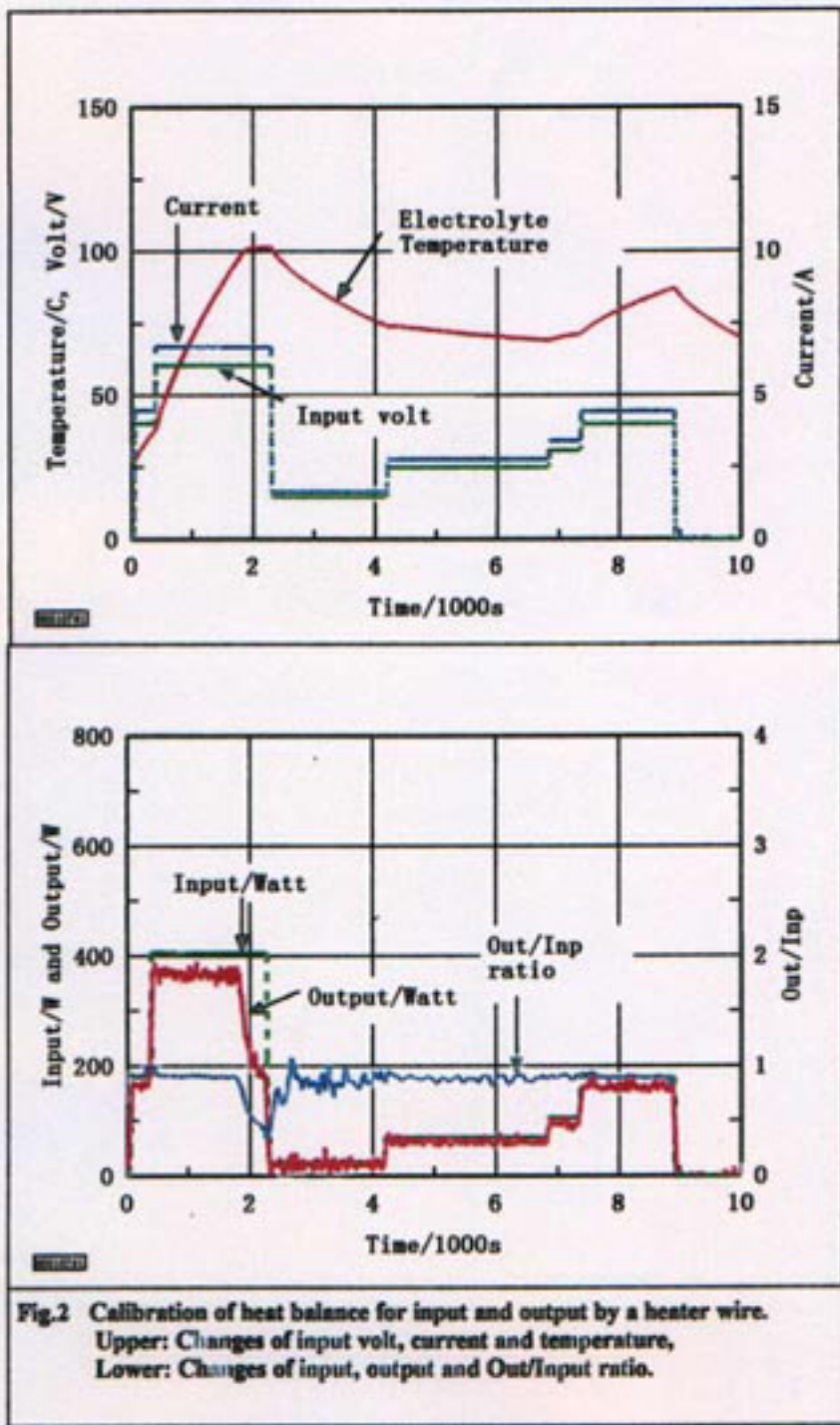
Dependence of heat release on cell temperature



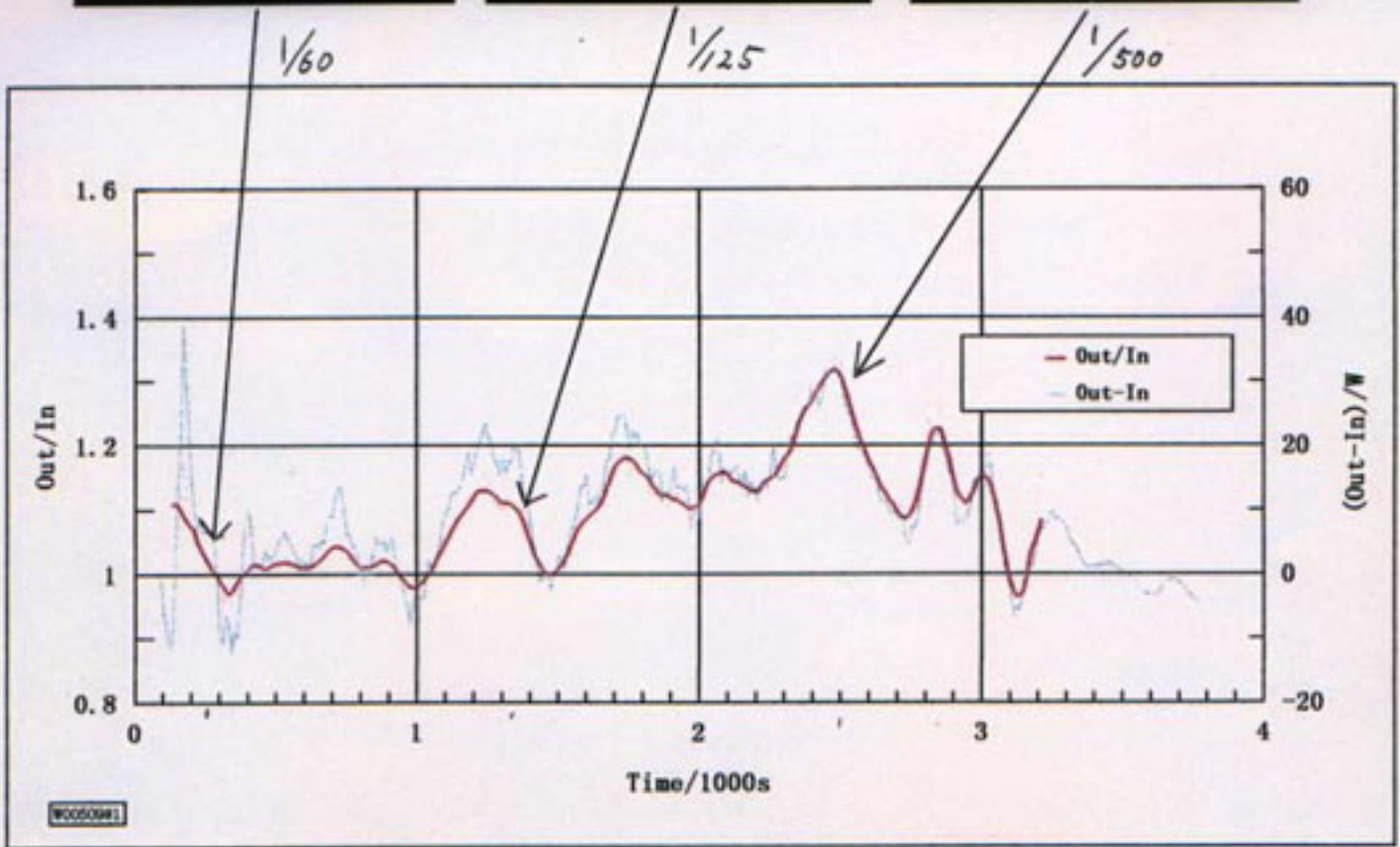


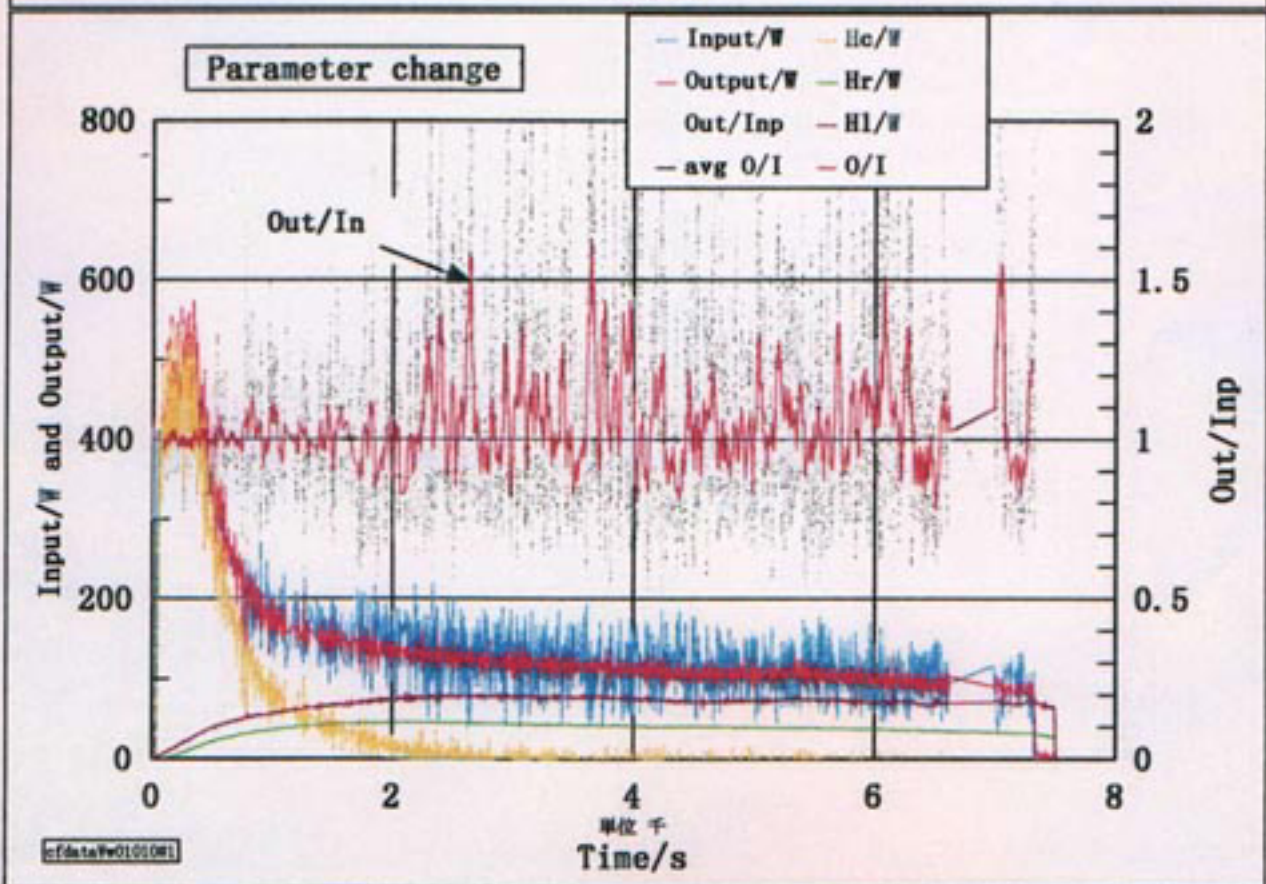
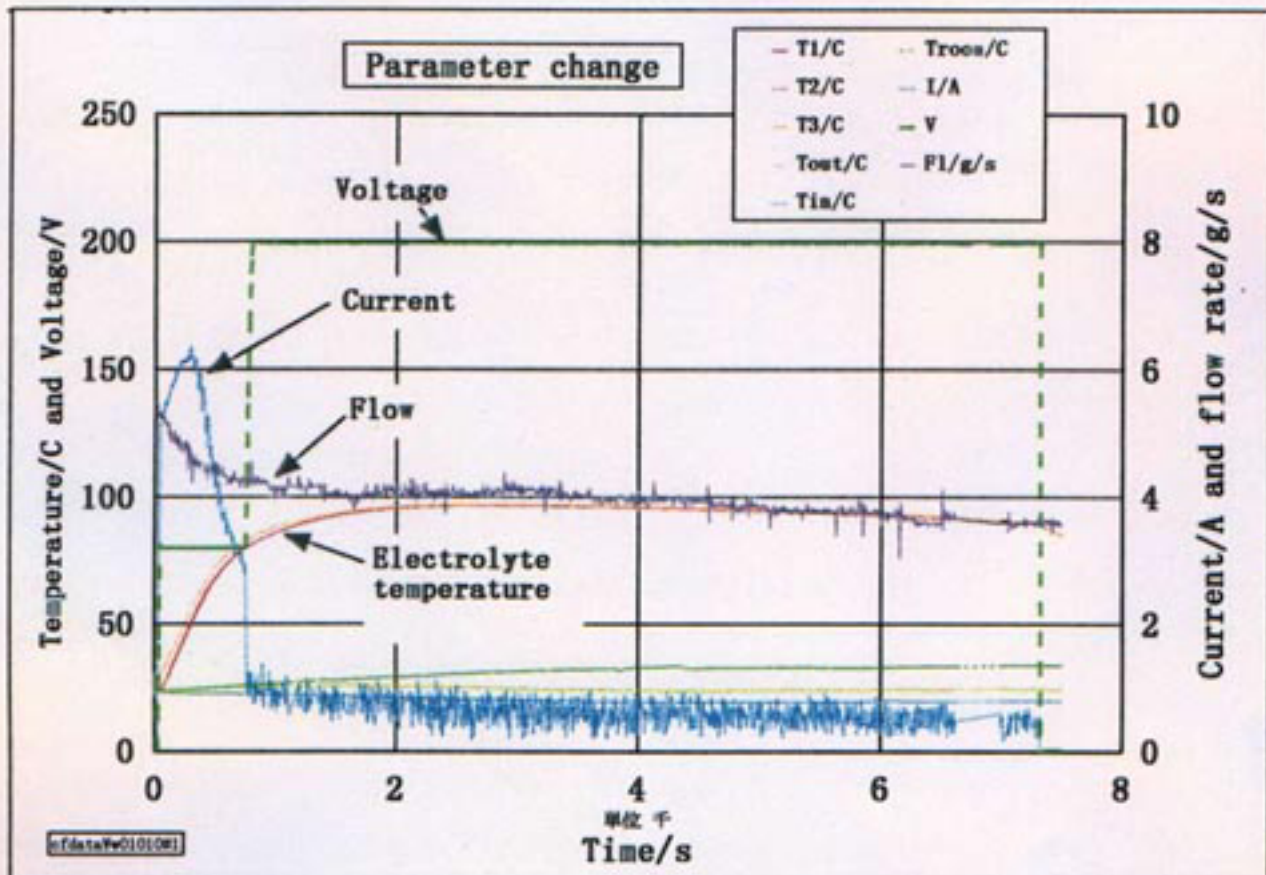
YcfdataYT1W99. wk4

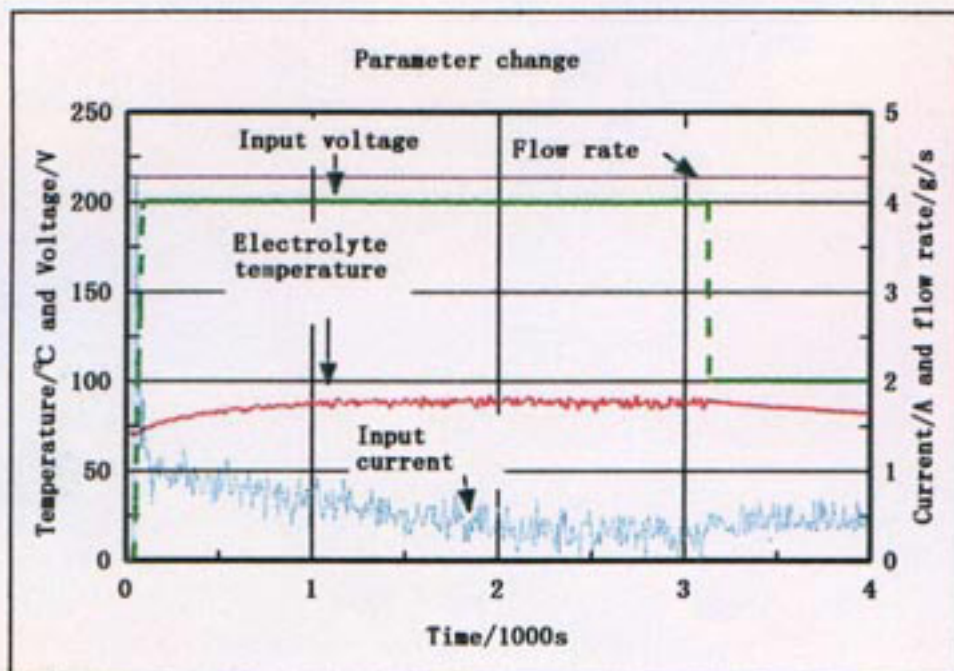
Current and Volt relation for plasma electrolysis



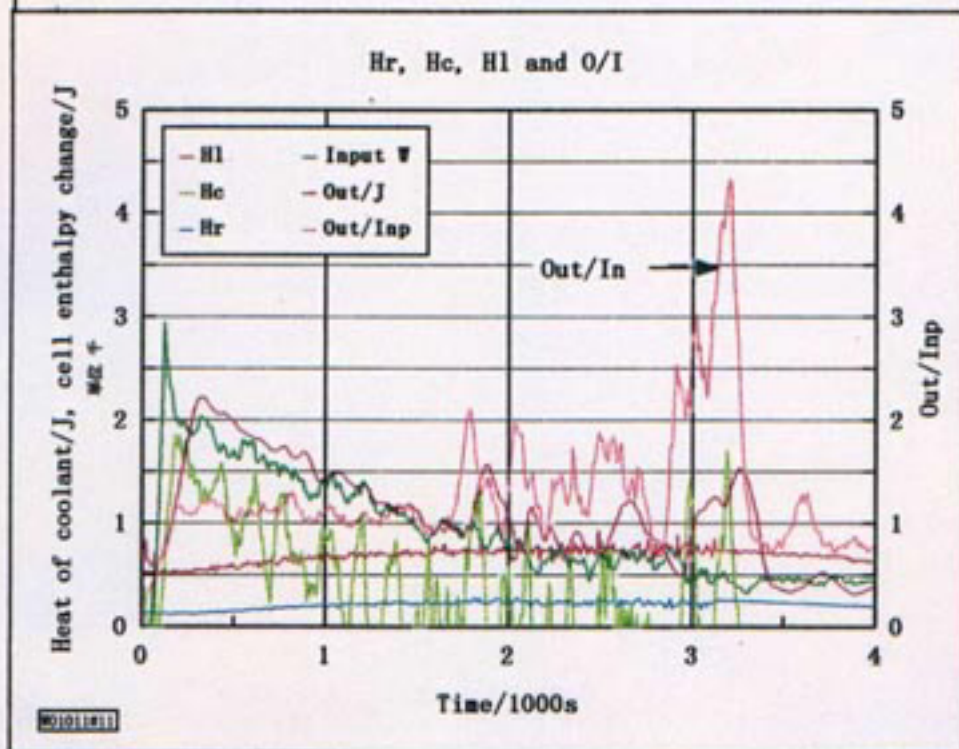
Changes of plasma discharge



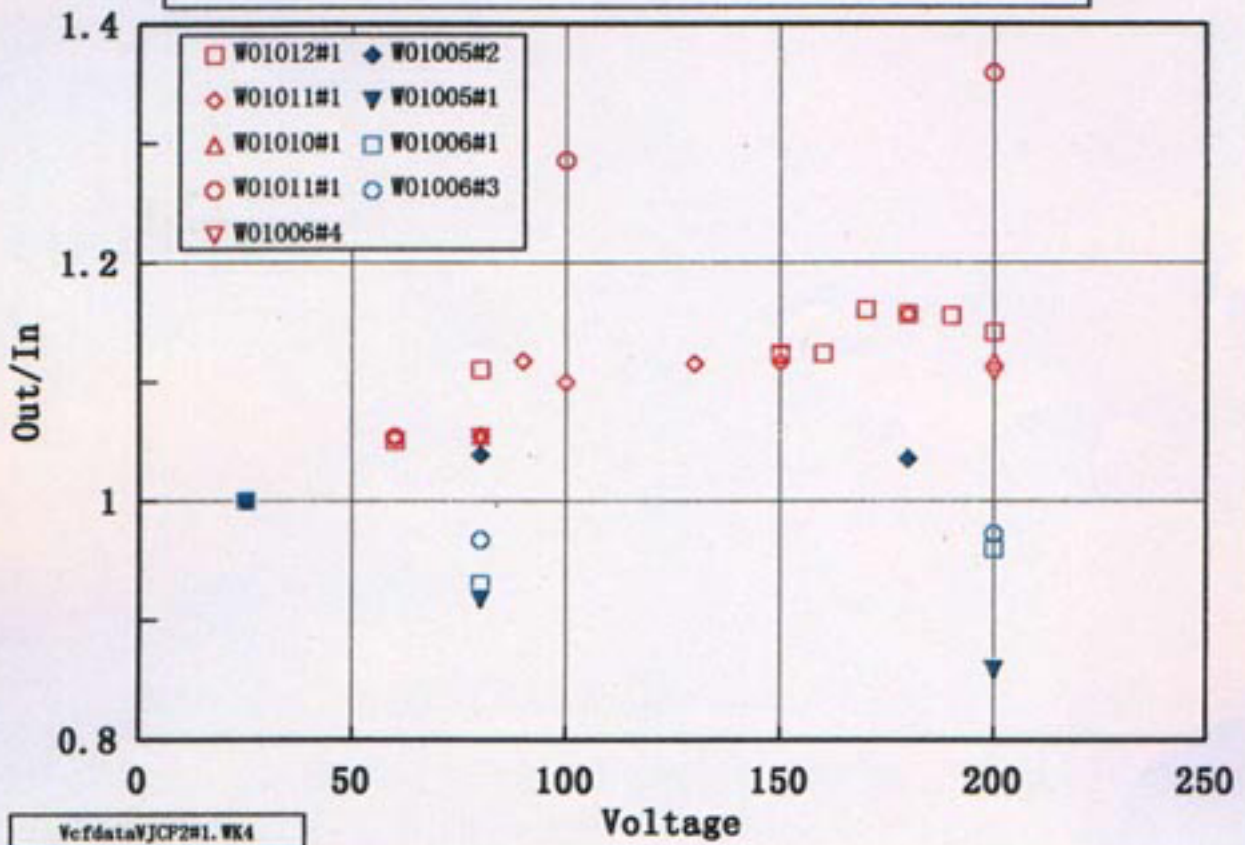




200V 1.29
 100V 1.22
 60V 1.00



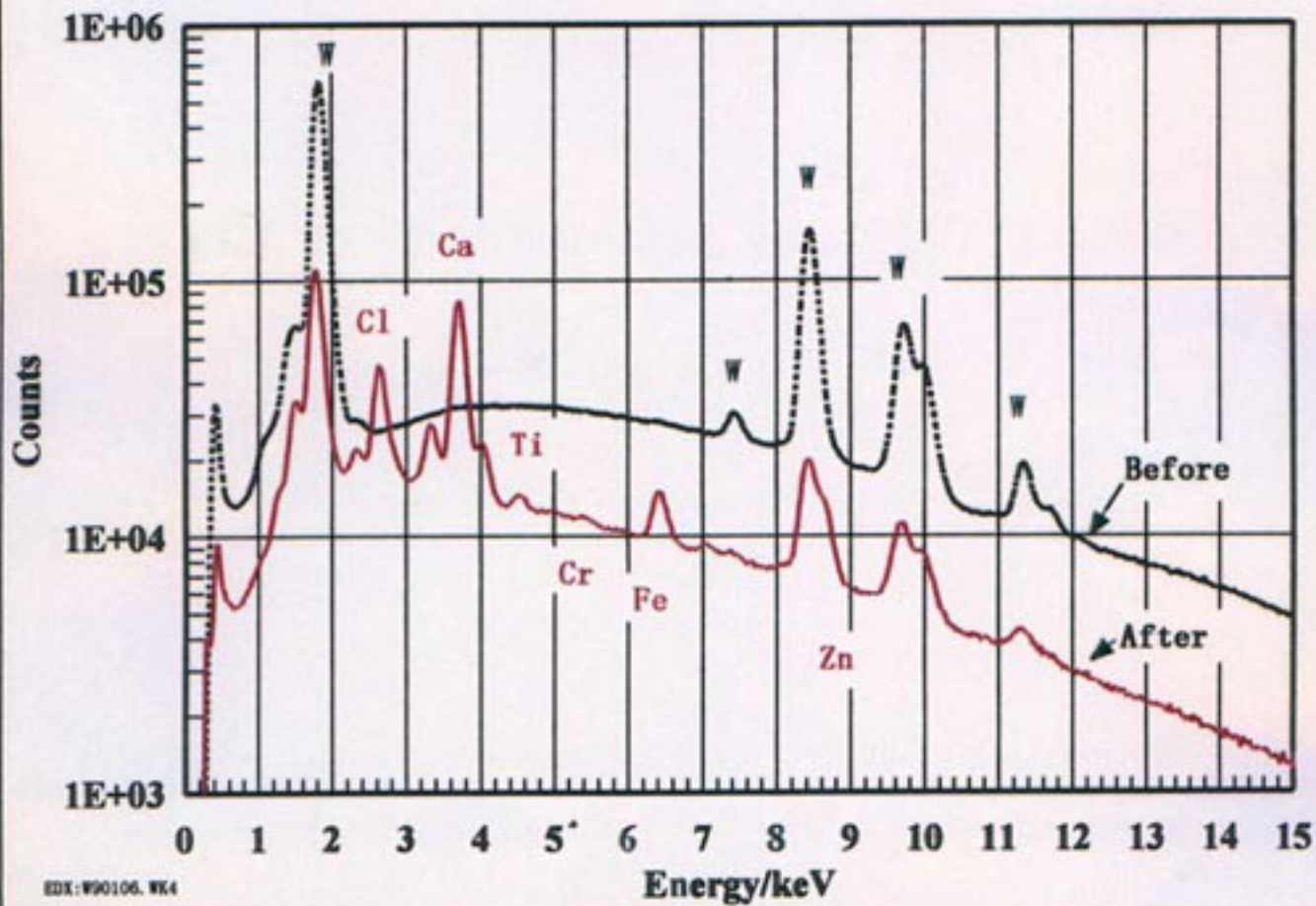
Out/In ratio change by input voltage



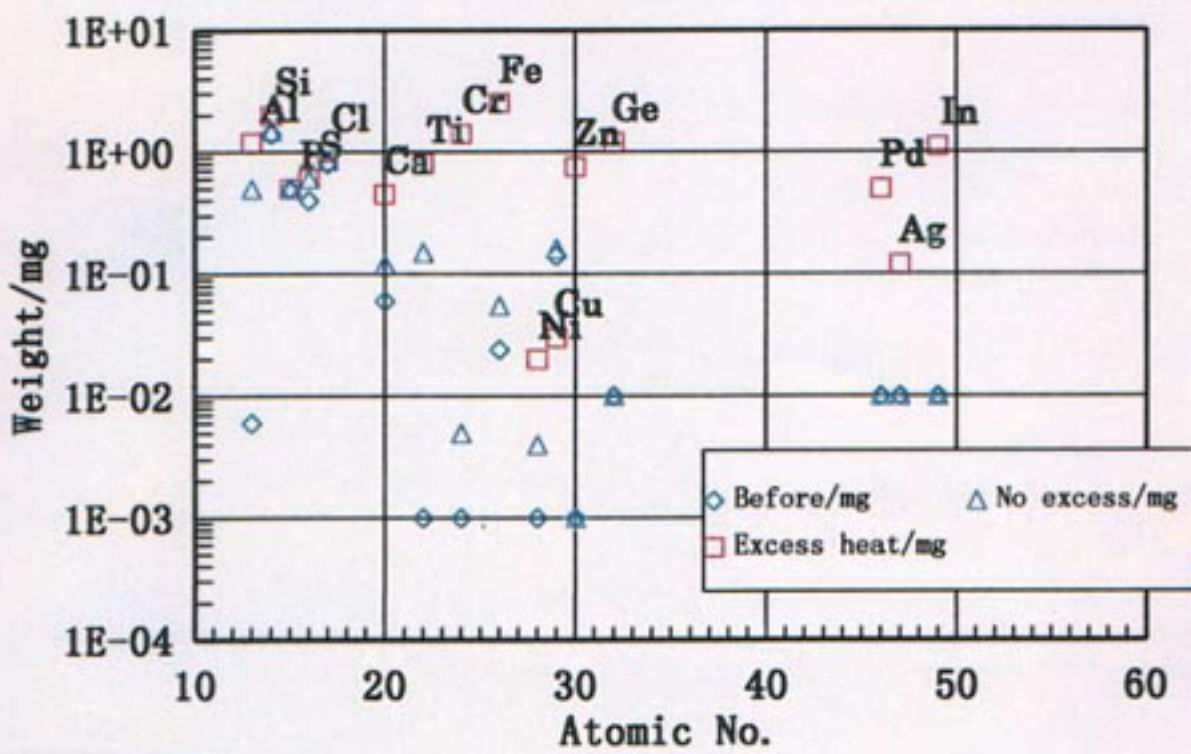
SEM image

1mm





**Energy spectra of EDX analysis for W electrode surfaces;
before and the after plasma electrolysis.**



Vcfdata\icpele20.wk4

Total amount of elements in the electrolyte and on the electrode for the case of before, excess heat and no excess heat evolved.

Summary

- Excess heat was confirmed during the plasma electrolysis of tungsten electrode.
- Many elements were detected on the electrode and in the electrolyte for the case of excess heat generated experiment.
- Experimental results can be well explained by the photo fission mechanism.